## Precise measurement of single molecular conductance using STM point contact method

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Charge transport in a single molecule is one of the central issues of current molecular electronics. A metal/single-molecule/metal junction shows a broad conductance distribution and stochastic switching caused by the varieties of molecular conformations and interface geometries. Despite the importance of the phenomena for both fundamental and practical advances, the relationship between the characteristics of single molecular conductance and molecular junction geometries has not yet been clarified until now. Here, we introduce our recent results on this issue.

We used  $\alpha, \omega$ -bis(acetylthio)terthiophene (3TS2) and  $\alpha, \omega$ -bis-(acetylseleno)terthiophene (3TSe2) molecules as samples to study the influence of junction geometry on the single molecular conductance. Isolated 3TS<sub>2</sub> or 3TSe<sub>2</sub> molecules embedded in a self-assembled monolayer (SAM) of octanethiol (C8) molecules were prepared. Measurements were performed by STM point contact method<sup>1)</sup> under UHV at 77K. Schematic illustration for the measurement conformation is shown in Fig. 1.

After forming a molecular junction by approaching the STM tip (Au) to a target molecule, several hundreds of I-V curves were measured. During the I-V curve measurement, the electrode gap space was changed to modulate the Molecular junction geometry. Figure 2(a) shows a sequence of conductance values calculated from the I-V curves. Conductance switching and rupture of molecular junction can be seen. Figure 2(b) shows a conductance histogram obtained for  $3TS_2$ , under a fixed gap space condition. Several conductance peaks (indicated by arrows) can be observed due to the variety of adsorption sites of molecule on an Au surface. While, only a large single conductance peak (~170nS) remains (Fig. 2(c)) when the data of conductance measured just before the rupture of molecular junction (last conductance step) is plotted. Since the binding force of a thiol-Au bonding is stronger than that of an Au-Au bonding, the extensional strain exerted on a molecule induces the migration of the Au atom bound to the molecule and adsorption site hops to Au top site. Therefore, a narrower distribution can be obtained for the conductance measured in the last step.

